

Mathematical-Chemical Investigation of some Carboxylic Acids

Kristine Giorgadze* and Mikheil Gverdtsiteli*

* I. Javakhishvili Tbilisi State University, Tbilisi

(Presented by Academy Member Shota Samsoniya)

ABSTRACT. Some carboxylic acids Mathematical-chemical investigation was carried out within the scope of quasi-ANB-matrices method. Three correlation equations were constructed and investigated. Correlations are satisfactory. Shannon's information entropies were calculated for these acids. © 2016 Bull. Georg. Natl. Acad. Sci.

Key words: carboxylic acids, quasi-ANB-matrix, correlation equation, information entropy.

Mathematical Chemistry is modern branch of theoretical chemistry. With application of categories of higher mathematics (groups, graphs, matrices and etc) it solves the classical problem of chemistry – correlation between the structure of compound and its static and dynamic (chemical transformations) properties. Graphs theory plays very important role in mathematical chemistry [1]. Contiguity matrices of molecular graphs and their modifications are widely used in mathematical chemistry for investigation of molecules and their transformations and ANB-matrix falls into this type [2-3].

The diagonal elements of ANB-matrix are the atomic numbers of the chemical elements, nondiagonal elements are the multiplicities of the chemical bonds.

For arbitrary XYV molecule, the ANB-matrix has the form:

$$\begin{pmatrix} Z_X & \Delta_{XY} & \Delta_{XV} \\ \Delta_{XY} & Z_Y & \Delta_{YV} \\ \Delta_{XV} & \Delta_{YV} & Z_V \end{pmatrix}$$

where: Z_X , Z_Y and Z_V are atomic numbers of X, Y, V chemical elements; Δ_{XY} , Δ_{XV} , Δ_{YV} are the multiplicities of X~Y, X~V, Y~V chemical bonds.

For large molecules calculations are very labour-consuming. Thus, the modernized form of ANB-matrix – quasi-ANB-matrix ($\tilde{A}\tilde{N}\tilde{B}$) was elaborated. Its diagonal elements are the sums of the atomic numbers of those chemical elements, which the structural fragments of the molecule contain; nondiagonal elements represent multiplicities of the chemical bonds between the structural fragments. Thus, $\tilde{A}\tilde{N}\tilde{B}$ -ma-

Table. $\lg(\Delta_{A\tilde{N}B})$, T_{boil} , d_4^{20} , S_{298}^0 and H_S for carboxylic acids

Carboxylic acids	$\lg(\Delta_{A\tilde{N}B})$	T_{boil} , °C	d_4^{20}	S_{298}^0 , cal/(mole·K)	H_S
CH ₃ COOH	2.31	118.1	1.049	38.20	1.5000
C ₂ H ₅ COOH	2.56	141.1	0.992	(40.80)	1.4360
C ₃ H ₇ COOH	2.76	163.5	0.959	54.1	1.370
C ₄ H ₉ COOH	2.88	187.0	0.942	62.1	1.3319
C ₅ H ₁₁ COOH	2.96	205.0	0.929	(67.4)	1.2955

trix is constructed on the basis of the molecular model. Decimal logarithm of the value of the determinant of $A\tilde{N}B$ -matrix is effective topologic index for construction of the correlation equations of „structure-properties” type.

Carboxylic acids were investigated within the scope of $A\tilde{N}B$ -matrices method. The simplest model was elaborated for them:



where: $R \equiv \text{CH}_3, \text{C}_2\text{H}_5, \text{C}_3\text{H}_7 \dots$; $Y \equiv \text{COOH}$.

Corresponding $A\tilde{N}B$ -matrix has the form:

$$\begin{vmatrix} Z_X & 1 \\ 1 & Z_Y \end{vmatrix}$$

In the table the values of $\lg(\Delta_{A\tilde{N}B})$, T_{boil} , d_4^{20} [4], S_{298}^0 [5] and H_S are represented for these acids.

Three correlation equations were constructed on

computer:

$$T_{\text{boil}} = 1.597 \cdot \lg(\Delta_{A\tilde{N}B}) - 267.7$$

$$d_4^{20} = -0.157 \cdot \lg(\Delta_{A\tilde{N}B}) + 1.394$$

$$S_{298}^0 = 66.7 \cdot \lg(\Delta_{A\tilde{N}B}) - 130.0$$

The correlations coefficient r is respectively equal to 0,981; 0,980; 0,983. Thus, in accordance with Jaffe's criterion [6], correlations are satisfactory. The values of S_{298}^0 for C₂H₅COOH and C₅H₁₁COOH are calculated theoretically, on the basis of corresponding correlation equation.

Shannon's information entropy is calculated by formula [7]:

$$H_S = -\sum p_i \log_2 p_i$$

where: p_i is the ability of some event (for example, to „choose” c-atom in the molecule).

For this acids the values of H_S were calculated and they are represented in the Table.

ორგანული და არაორგანული ქიმია

ზოგიერთი კარბონმჟავას მათემატიკურ-ქიმიური გამოკვლევა

ქ. გიორგაძე*, მ. გვერდსითელი*

* თანე ჯგუზიშვილის სახ. თბილისის სახელმწიფო უნივერსიტეტი

(წარმოდგენილია აკადემიის წევრის შ. სამსონიას მიერ)

ჩატარებულია ზოგიერთი კარბონმჟავას მათემატიკურ-ქიმიური გამოკვლევა კვაზი-ანბ-მატრიცების მეთოდის ფარგლებში. აგებულია სამი კორელაციური განტოლება. კორელაციები დამაკმაყოფილებელია. ამ მჟავებისათვის გამოთვლილია შენონის ინფორმაციის ენტროპიის მნიშვნელობები.

REFERENCES

1. Rouvray P.H. (1983) Chemical application of topology and graph theory /Ed. T. Balaban. Amsterdam.
2. Kupatadze K., Lobzhanidze T., Gverdtsiteli M. (2007) Algebraic-chemical investigation of some organic molecules and their transformations. Tbilisi.
3. Gverdtsiteli M., Gamziani G., Gverdtsiteli I. (1996) The contiguity matrices of molecular graphs and their modifications (Tbilisi).
4. Nesmianov A.N., Nesmianov N.A. (1969) Nachala organicheskoi khimii (The Basis of Organic Chemistry). Moscow (in Russian).
5. Stull D., Westrum E., Sinkl G. (1971) Khimicheskaia termodinamiĉa organicheskikh soedinenii (The Chemical Thermodynamics of Organic Compounds). Moscow (in Russian).
6. Gverdtsiteli M. (1971) Physical Organic Chemistry. p.19. Tbilisi (in Georgian).
7. Djdanov Yu. (1971) Teoriia stroeniia organicheskikh soedinenii. Moscow (in Russian).

Received February, 2016